

a)  $\text{Au}(\text{Br})_2$  – gold (II) bromide

b)  $\text{CdCl}_2$  – cadmium (II) chloride

c)  $\text{Na}_2\text{O}_2$  – sodium peroxide

d)  $\text{SO}_4$  – sulfate

e)  $\text{Zn}_3\text{P}_2$  – trizinc diphosphide

f)  $\text{BeHPO}_4$  – beryllium hydrogen phosphate

g)  $\text{HCl}$  – hydrochloric acid

h)  $\text{HgNO}_3$  – mercury (I) nitrate

i)  $\text{Ni}_2\text{SO}_3$  – nickel sulfite

j)  $\text{TiOH}_3$  – titanium (III) hydroxide

k) sulfuric acid –  $\text{H}_2\text{SO}_3(\text{aq})$

l)  $\text{CoCO}_3$  – copper (II) carbonate